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Silvia E. Ramos et al. RBQ Vol.39, No.4, pp. 94-105, 2022

Received 04 22 2022 Accepted 10 24 2022 Published 10 30 2022; DOI:10.34098/2078-3949.39.4.1 Determinación de parámetros de secado por aspersión para la obtención de extracto seco rico en saponinas de residuos de escarificado de quinua

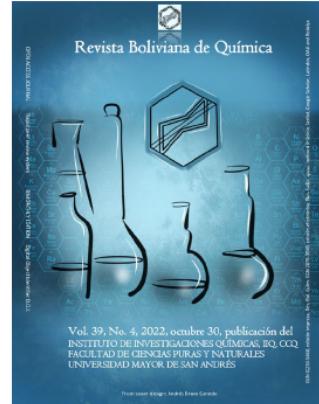


DETERMINACIÓN DE PARÁMETROS DE SECADO POR ASPERSIÓN PARA LA OBTENCIÓN DE EXTRACTO SECO RICO EN SAPONINAS DE RESIDUOS DE ESCARIFICADO DE QUINUA

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Full original article

Peer-reviewed

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Keywords: Spray Drying, Saponins, Hydroalcoholic Extract, Chenopodium quinoa Residues

Palabras clave: Secado por Aspersión, Saponinas, Extracto Hidroalcohólico, Residuos de Chenopodium quinoa

ABSTRACT

Determination of spray-drying parameters to obtain a dry extract rich in saponins from quinoa scarified residues. Quinoa scarification residues are generated in large quantities in Bolivia and have a high content of saponins, compounds with various properties that give them a potential use in cosmetics, agriculture, health and others. For this reason, it is important to have methods to obtain dry extracts rich in saponins from these residues. In the present work, we determined some parameters for spray-drying the aqueous residual extract from a hydroalcoholic extraction process: EtOH:H₂O (50:50) for 72 h, followed by removal of EtOH by rota-evaporation. For this, a 2³ factorial experimental design was used, taking as independent variables: the concentration of the residual aqueous extract, the drying temperature and the speed of the air flow. It was determined that the significant variables to obtain a higher extraction yield in the spray drying process are: the drying temperature (100°C) and the speed of the air flow (4192 rpm), with a feed flow of 8.0mL/min. The dry product obtained has a yield of 27.6% respect of quinoa residues and



68.73% of saponins. The spray-drying technique presented some advantages over the lyophilisation, such as the higher percentage of saponins in the dry extract and the shorter time used for drying.

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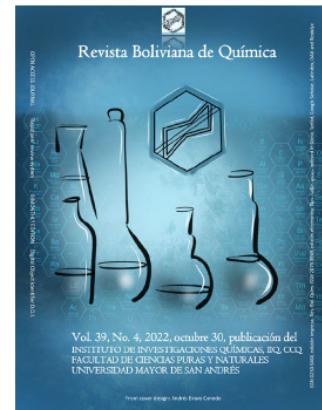
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Mesoporous silicoaluminate materials (MCM-41, SBA-15 and MCF) by atrane route for cobalt catalyst

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Revista Boliviana de Química39(4), 106-125, Sep./Oct. 2022
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DOI: 10.34098/2078-3949.39.4.2**MESOPOROUS SILICOALUMINATE MATERIALS (MCM-41, SBA-15 AND MCF) BY ATRANE ROUTE FOR COBALT CATALYST***Full original article**Peer-reviewed*Mauricio Claure Zeballos^{1,2,*}, Fatima L. Pardo Tarifa¹, Luis G. Lopez N.^{1,2}, Saúl Cabrera M.^{1,2†}¹ Laboratorio de Ciencia de Materiales, Instituto de Investigaciones Químicas IIQ, Department of Chemical Sciences, School of Pure and Natural Sciences FCPN, Universidad Mayor de San Andrés UMSA, P.O. Box 303, Calle Andrés Bello s/n-Edificio IIQ, Ciudad Universitaria Cota Cota, Phone +59122795878, La Paz, Bolivia, <http://cienciasquimicas.umsa.bo/>² Instituto del Gas Natural IGN, Universidad Mayor de San Andrés UMSA, P.O. Box 303, Calle Andrés Bello s/n-Edificio IIQ, Ciudad Universitaria Cota Cota, Phone +59122772269, La Paz, Bolivia. sicyt.umsa.bo/unidades/contactoInstituto/103.**Keywords:** Mesoporous silicoaluminate, MCM-41, SBA-15, MCF, catalyst support, Cobalt Catalyst**Palabras clave:** Silicoaluminatos mesoporosos, MCM-41, SBA-15, MCF, soporte de catalizador, catalizador de cobalto**ABSTRACT**

In this work, the synthesis of mesoporous silicoaluminum supports synthesized by the atrane route was performed. The obtained supports presented high homogeneity of the aluminum dispersion, high surface area and narrow pore size distribution. The synthesized mesoporous supports were: MCM-41, SBA-15 and MCF with 10% of Al_2O_3 in the matrix of SiO_2 . Twelve percent of cobalt was added to these supports by the incipient wetness impregnation method. These materials were characterized by adsorption of Nitrogen (BET-BJH), Scanning Microscopy Electron (SEM), X-Ray Diffraction (XRD), H_2 Temperature-Programmed Reduction (TPR) and NH_3 Temperature-Programmed Desorption (TPD). According to their structural properties of these catalysts, a promising application in the Fischer-Tropsch syntheses (FTS) is identified.



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César A. Andrade-Tacca et al. RBQ Vol.39, No.4, pp. 126-138, 2022

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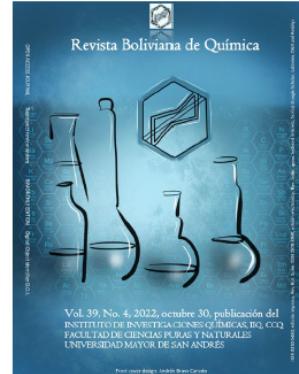
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Electrodeposición de dióxido de manganeso
electrolítico en sustratos de carbón a partir de
lixiviados de pilas alcalinas agotadas**ELECTRODEPOSICIÓN DE
DIÓXIDO DE MANGANESO
ELECTROLÍTICO EN SUSTRATOS DE
CARBÓN A PARTIR DE LIXIVIADOS
DE PILAS ALCALINAS AGOTADAS**

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Revista Boliviana de Química39(4), 126-138, Sep./Oct. 2022
Bolivian Journal of Chemistry
DOI: 10.34098/2078-3949.39.4.3*Full original article**Peer-reviewed*César A. Andrade-Tacca^{1,2,*}, Raghyth L. Patiño Patiño³, Edwin Urday Urday⁴¹ Escuela Profesional de Ingeniería Ambiental, Facultad de Ingeniería de Procesos FIP, Universidad Nacional de San Agustín de Arequipa UNSA, Av. Independencia s/n Pab. Antonio Raimondi 4to piso, Área de Ingenierías, Tel. +5154201723, Arequipa, Perú, ambiental@unsa.edu.pe, <http://fip.unsa.edu.pe/ingambiental/>² Instituto de Investigación de Ciencia y Gestión Ambiental "Antonio Brack" ICIGA-UNSA, Universidad Nacional de San Agustín de Arequipa UNSA, San Agustín 104, 2do Piso, Tel. 5154231731, Arequipa Perú, icigaunsaambiental@unsa.edu.pe³ Escuela Profesional de Ingeniería Metalúrgica, Facultad de Ingeniería de Procesos FIP, Universidad Nacional de San Agustín de Arequipa UNSA, Av. Independencia s/n-Pab. Ing. Metalúrgica, Tel. +5154282813, Arequipa, Perú, metalurgia@unsa.edu.pe, <http://fip.unsa.edu.pe/ingmetalurgica/>⁴ Escuela Profesional de Ingeniería de Materiales, Facultad de Ingeniería de Procesos FIP, Universidad Nacional San Agustín de Arequipa UNSA, Av. Independencia s/n-Pab. Ing. Materiales, Tel. +5154200037, Arequipa, Perú, materiales@unsa.edu.pe, <http://fip.unsa.edu.pe/ingmateriales/>**Keywords:** Alkaline batteries, Manganese dioxide, Leaching, Carbon cloth, Carbon felt**Palabras clave:** Pilas alcalinas, Dióxido de manganeso, Lixiviación, Telas de carbón, Filtros de carbón**ABSTRACT**

Electrodeposition of electrolytic manganese dioxide on carbon substrates from leachates from exhausted alkaline batteries. In the alkaline batteries of Zn/MnO₂, the electrolytic manganese dioxide (EMD) is used as cathode and constitutes a very important part of household waste which contributes with heavy metals to the solid urban waste. This study describes the recovery of EMD by electro-oxidation on carbon cloth in acidic leached liquors from used



alkaline batteries. The maximum extraction yield for Mn and Zn were 43.68% (0.744 molL^{-1}) and 56.11% (0.7321 molL^{-1}) respectively. The voltammograms of a vitreous carbon electrode for EMD deposition from leached liquors were obtained between 0.0 and 1.4V vs ECS, and exhibit an anodic peak at 1.2V vs ECS. In addition, a cathodic peak appears at 0.97V vs ECS. The x-ray diffraction reveals the presence of nsutite ($\gamma\text{-MnO}_2$) and manganite ($\gamma\text{-MnOOH}$) for all deposits, and the morphology of the deposits was observed by SEM.

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Pictorial mechanistic views on the chemical history of withaferin A: the structural determination of the side chain, The Organic Chemistry Notebook Series, a Didactical Approach, N° 16



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PICTORIAL MECHANISTIC VIEWS ON THE CHEMICAL HISTORY OF WITHAFERIN A: THE STRUCTURAL DETERMINATION OF THE SIDE CHAIN, THE ORGANIC CHEMISTRY NOTEBOOK SERIES, A DIDACTICAL APPROACH, N° 16

Analytical review

Peer-reviewed

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Keywords: *Organic Chemistry, Withaferin A, Withania somnifera, Side chain, Structure determination, Mechanisms of Reactions.*

Palabras clave: *Química orgánica, Withaferin A, Withania somnifera, Cadena lateral, Determinación de estructuras, Mecanismos de reacciones.*

ABSTRACT

This article is an approach to the historical development of the structural elucidation of organic natural substances that began with chemical degradations, isotopic labeling processes and sometimes synthesis, all chemical reactions. We used the work of Lavie et al. in: "Constituents of *Withania somnifera* Dun. III. The side chain of Withaferin A" as the study-object, complemented by other bibliographic material, and we proposed mechanistic routes of each reaction used by the authors. The objective was to explain these processes from the didactic point of view of reaction mechanisms.